

Predicting sensory evaluation indices of Cheddar cheese texture by fluorescence fingerprint measurement coupled with an optical fibre

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21 ABSTRACT

22	A fluorescence fingerprint (FF) was used to develop a quick and non-contact practical method for
23	predicting the sensory evaluation index of cheese texture (cheese body measurement). A partial
24	least-squares (PLS) model was constructed from FF data and cheese body measurements of Cheddar
25	cheeses. The cheese body measurement was successfully predicted by the PLS model with a coefficient of
26	determination for calibration of 0.800. Notably, the reproducibility of the prediction value and the model
27	accuracy were comparable to those of a conventional FF model despite the non-contact measurement. By
28	exploring the variable importance in projection (VIP) and selectivity ratio (SR) of the PLS model, the
29	fluorescence likely corresponding to oxidized lipids, Maillard reaction products, and compounds of
30	proteins and amino acids with oxidized lipids was found to increase in intensity with the progress of
31	ripening. This suggests that the fluorescence of these compounds contributes to the accuracy of the PLS
32	model.
33	

# 35 1. Introduction

36	The composition (fat, moisture, and intact casein) and maturity of natural cheese as a raw material for
37	processed cheese has a significant impact on the product quality and stability (Guinee, Caric, & Kalab,
38	2004; Kapoor & Metzger, 2008; Tamime, 2011). Generally, young cheese is used to produce hard
39	processed cheese, whereas aged cheese is used to produce soft processed cheese. For example, block-type
40	processed cheeses that are easy to slice and are elastic require a young cheese, whereas spread-type
41	processed cheeses are primarily based on cheese with medium maturity. However, as natural cheese starts
42	to mature immediately after manufacturing, the maturity of individual natural cheeses used to make
43	processed cheese will differ. Therefore, to achieve products with sufficient uniformity, the blending ratio
44	must be adjusted daily.
45	Processed cheese has numerous advantages including 1) a long shelf-life; 2) the characteristics of
46	multiple natural cheeses through which, with additional seasoning, variations in the taste can be enhanced;
47	3) a mild taste, making it palatable for children and people who may not be familiar with natural cheese;
48	4) an easily modified shape and readily adjustable texture (elasticity, hardness, spreadability, and ease of
49	slicing); and 5) high suitability for cooking (flow, ease of browning, and viscosity).
50	Assessment of the texture of cheese based on detailed measurements represents an active area of
51	research. In particular, the relationship between physical properties and texture evaluated through
52	instrumental measurements and sensory evaluation (Drake, Gerard, Truong, & Daubert, 1999; Foegeding,
53	Brown, Drake, & Daubert, 2003; Guinee & Kilcawley, 2004; Lee, Imoto, & Rha, 1978; Morita et al.,
54	2015; O'Callaghan & Guinee, 2004; Xiong, Meullenet, Hankins, & Chung, 2002) constitutes a primary
55	area of interest. However, only a few measurement methods are actively used by processed cheese
56	manufacturers to evaluate natural cheese as raw materials. An example of such methods currently on the
57	market includes "Caseus Pro <sup>TM</sup> ," (Gold Peg International, Inc., Braeside, Australia). This is because the

59	pretreatment process for evaluation and instrumental measurement is complex, the measurement results
60	cannot be directly related to traditional sensory evaluation results, and the number of samples required for
61	appropriate evaluation is excessively large. Therefore, in general, sensory evaluation approaches are faster.
62	Individual processed cheese manufacturers continue to use traditional sensory evaluation or only employ
63	instrumental measurement data directly, with each company using its own methods. For example, an index
64	termed "cheese body measurement" is used for the sensory evaluation of physical properties. The cheese
65	body measurement is an index of the physical maturity determined by trained experts that is based on the
66	physical sensation of crumbling a piece of cheese between the fingers. This index varies between 1 and 10,
67	where 10 points indicates a strong body (young texture without maturing) whereas 1 point indicates a
68	weak body (matured texture), and is likely to vary significantly among cheese manufacturers
69	(Mizuno & Ichihashi, 2007; Muir, 2010; Nakazawa & Hosono, 1989).
70	Alternatively, methods have been developed that utilize mid-infrared light, near-infrared light,
71	fluorescence, and Raman spectroscopy to predict the components and physical properties of food, which
72	are used as quick measurement methods (Nawrocka & Lamorska, 2013). As fluorescence and
73	near-infrared spectra can be measured without sample pretreatment, these spectroscopic methods exhibit
74	high potential for industrial use. In particular, fluorescence spectroscopy measures the emission spectrum
75	instead of the absorption spectrum and demonstrates high sensitivity compared to that of near-infrared
76	light (Karoui, Mazerolles, & Dufour, 2003; Kulmyrzaev, Karoui, De Baerdemaeker, & Dufour, 2007).
77	Fluorescent compounds are sensitive to their surrounding environment (e.g., temperature, ionic
78	concentration, pH, and polarity of solution); moreover, fluorescence spectroscopy measurements can be
79	rapidly performed (Dufour, 2011). In addition, some investigations have been performed involving the
80	measurement of cheese using fluorescence spectroscopy (Andersen & Mortensen, 2008; Christensen,
81	Nørgaard, Bro, & Engelsen, 2006; Dufour, 2011; Karoui & Blecker, 2011). Furthermore, with the
82	development of information processing technologies in recent years, fluorescence spectroscopy

measurements allow the rapid acquisition of emission spectra from excitation light with a continuous
wavelength, such as in the form of an excitation-emission matrix (EEM), also known as a fluorescence
fingerprint (FF).

86 The FF has an advantage over conventional fluorescence spectra because it includes emission spectra 87 excited at many different excitation wavelengths. Thus, FF yields much more information as compared 88 with the study of conventional single or double excitation wavelengths, and has the potential to estimate 89 phenomena with high accuracy. This approach is used as a measurement method by which a large amount 90 of information can be acquired and processed (Airado-Rodríguez, Galeano-Díaz, Durán-Merás, & Wold, 91 2009). A FF targets not only the peak intensity of the fluorescence signal but also other wavelength ranges 92 in which the fluorescence signal is weak. A model can then be constructed that only extracts necessary 93 information. There are several instruments that can derive predictions based on fluorescence and a FF. 94 Although a few analytical cases have been examined and numerous discussions have been conducted 95 regarding the predictive accuracy obtained using these techniques, to our knowledge no reports are yet 96 available of a detailed prediction model based on the contributing wavelength range or other factors 97 (Lacotte et al., 2015; Liu, Sajith Babu, Coutouly, Allouche, & Amamcharla, 2016). However, there have 98been some cases of FFs being applied to study food (Sádecká & Tóthová, 2007) including dairy products 99 containing cheese (Andersen & Mortensen, 2008; Boubellouta & Dufour, 2008; Christensen, Povlsen, & 100 Sørensen, 2003). 101 Although there have been some studies that predicted cheese texture and physical properties using

102 fluorescence spectroscopy (Garimella Purna, Prow, & Metzger, 2005; Karoui et al., 2003; Karoui &

103 Dufour, 2006; Kulmyrzaev et al., 2005; Lebecque, Laguet, Devaux, & Dufour, 2001; Ozbekova &

104 Kulmyrzaev, 2017), their predictions were based on the fluorescence of a limited number of entities such

105 as tryptophan and vitamin A. Moreover, the emission spectra were measured by using only single or

106 double excitation wavelengths, with the primary purpose being to analyze the correlation between the

 $\mathbf{5}$ 

107	peak intensity and components or textures. Recently, Kokawa et al. (2015) showed that a FF can predict
108	indices of cheese maturation. They predicted the maturation time and some chemical analysis values,
109	such as the proteolysis index (the ratio of water-soluble nitrogen content to total nitrogen content) and
110	total free amino acid content, from FFs. They also demonstrated that the FF constitutes an effective tool
111	for capturing changes in cheese through maturation. However, their method of measurement still requires
112	the preparation of samples with a shape that fits the cell of the fluorescence spectrophotometer. In
113	addition, there have been no studies where a FF has been used to predict the cheese body measurement,
114	the index of sensory evaluation for cheese texture, for the raw materials of processed cheese.
115	In this study, our aim was to develop a rapid method for predicting the cheese body measurement
116	using a FF, regardless of the experience level of the user. Then, we attempted to develop a method for
117	non-contact prediction of the cheese body measurement by exploiting the optical fiber unit in a
118	fluorescence spectrophotometer. In consideration of the practicality for industrial use, we have chosen a
119	non-contact measurement using a fiber optics unit to eliminate the pretreatment of the sample. Although
120	some reports are available in areas other than dairy research, no examples yet exist of application of this
121	method to cheese texture. By removing the sample pretreatment step, measurement of the FF was easily
122	achieved. Non-contact measurement of changes in the cheese with maturation will thus be faster than
123	traditional methods, rendering this method useful for processed cheese manufacturers.
194	

### 125 **2.** Materials and Methods

126 2.1. Cheese samples

127 Ten Australian Cheddar cheese samples were used for measurements. We performed three
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128 measurements and evaluations over time and obtained the results for a total of 30 samples. These cheese

129 samples were matured at -2 °C, 5 °C, 10 °C, and 15 °C to achieve differences in the cheese body

- 130 measurement. Each cheese sample was divided into blocks for FF measurement and cheese body
- 131 measurement evaluation. Cheese samples were also analyzed for moisture (ISO/IDF 4: 2004a), fat
- 132 (ISO/IDF 5: 2004b), and protein (ISO/IDF 185: 2002). The pH was measured by insertion of a pH probe
- 133 (D-54, Horiba, Kyoto, Japan) into the ground cheese sample.
- 134

## 135 2.2. Evaluation of cheese body measurement

136 The cheese body measurement was evaluated by five trained experts in consultation after adjusting

- 137 the temperature of the cheese to 5 °C. The maximum score for the cheese body measurement was 10,
- 138 corresponding to the strongest texture (young texture), with a score of 1 corresponding to the weakest
- 139 texture (matured texture). The cheese body measurement and FF were measured on the same day or on

140 consecutive days so that there was no difference in maturation.

141

### 142 2.3. FF measurement

143 To measure the FF of cheese, we used a fluorescence spectrophotometer (F-7000, Hitachi

144 High-Technologies Corporation, Tokyo, Japan) and an optical fiber unit (5J0-0114-F-7000, Hitachi High-

- 145 Technologies) (Mita Mala et al., 2016). The cheese samples subjected to FF measurement were adjusted
- 146 to 20 °C and cut into blocks of 3 cm  $\times$  3 cm  $\times$  6 cm. The side of the cheese samples was selected
- 147 randomly and was cut immediately prior to FF measurement, and measurements were taken on a fresh
- 148 surface each time. The samples were then placed on a stage with a fixed fiber unit. During measurement,

149external light was excluded from the sample using an enclosure. The measurement conditions for the FF 150were an excitation wavelength of 200-500 nm, an emission wavelength of 200-800 nm, and a wavelength 151interval of 5 nm. The slit width of the monochromator was 10 nm for both excitation and emission wavelengths. A photomultiplier voltage of 500 V was used and the scan speed was 60,000 nm min<sup>-1</sup>. The 152153distance between the probe tip of the fiber unit and the cheese sample was 5 mm, and the measurement 154was made without contact. Three replicates were measured for each cheese sample. 155

### 1562.4. PLS model for prediction of cheese body measurement

157 To predict the cheese body measurement from the FF measurements, a partial least-squares (PLS)

158regression model was constructed. For the multivariate analysis, MATLAB (R2016a) software

159(MathWorks Inc., Natick, MA, USA) and PLS Toolbox version 8.1.1 (Eigenvector Inc., Manson, WA

160USA) were used. FF data were preprocessed using the method of Yoshimura et al. (2014). First, as

161fluorescence constitutes an emission with wavelengths longer than the excitation wavelength, all data with

162emission wavelengths shorter than the excitation wavelengths were removed. Next, the primary light and

163the secondary and tertiary scattered lights, which comprise lights with the emission wavelengths at twice

164and three times the excitation wavelength (Fujita, Tsuta, Kokawa, & Sugiyama, 2010), respectively, were

165removed. Finally, excitation wavelengths shorter than 230 nm were removed because they contained

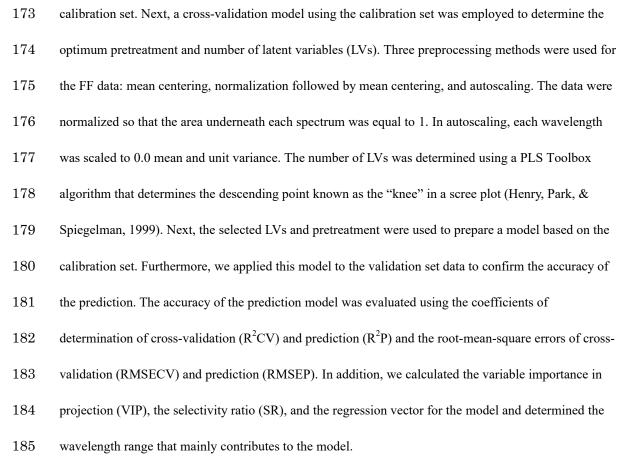
166significant noise. The remaining data were a combination of 3,869 excitation wavelengths and emission

167 wavelengths.

168 In total, we performed PLS regression analysis using 90 (three replicates  $\times$  10 samples  $\times$  three

169 evaluation periods) evaluation results for FF data. The cheese body measurement was measured for 30

- 170samples (10 samples × three evaluation periods). First, the data were divided into calibration and
- 171validation sets (2:1). Specifically, the samples were aligned in order of measurement and measurements
- 172for every third sample were allocated to the validation set and those for the rest were allocated to the



### 187 **3. Results and Discussion**

188 3.1. Fluorescence fingerprints and cheese compositions

189 Figure 1 shows the mean FF for the cheese samples. An emission peak was observed in the excitation

190 wavelength range of 290–305 nm and the emission wavelength range of 320–350 nm. This peak had a

191 fluorescence intensity of 6131 (a.u.) and has been determined to correspond to aromatic amino acids

192 including tryptophan (Andersen, Vishart, & Holm, 2005; Andersen & Mortensen, 2008; Mazerolles et al.,

- 193 2001). In addition, the intensity was high (1201 (a.u.)) at the excitation wavelength of 320 nm and
- 194 emission wavelength of 400 nm, which are likely the wavelengths of vitamin A. We focused on the peak
- 195 at ex320/em400 because it has been examined in many previous reports. Fluorescence measurements for
- 196 milk (Boubellouta & Dufour, 2008; Dufour & Riaublanc, 1997), soft cheese (Herbert et al., 2000),
- 197 semi-hard cheese (Karoui, Dufour, & De Baerdemaeker, 2006), and processed cheese (Christensen et al.,
- 198 2003) have been previously reported. The peak at the excitation wavelength of 300 nm and the emission
- 199 wavelength of 680 nm represents the second-order light of the excitation wavelength of 300 nm and the
- 200 emission wavelength of 335 nm, which appears owing to the light dispersion mechanism of the
- 201 monochrometer (i.e., the same mechanism that creates the second-order scattering light). In the FF data,
- similar peaks were detected to those in Kokawa et al. (2015); thus, we confirmed that a non-contact FF
- 203 could be obtained for cheese using an optical fiber.

Table 1 shows the age, composition and pH of 10 sample cheeses. It was confirmed that there was little difference in composition and pH between samples.

206

### 207 3.2. PLS prediction model

Table 2 shows the LVs and prediction accuracy of the PLS regression model for each pretreatment. Before performing the PLS regression analysis, four pretreatments were evaluated to optimize the model in terms of R<sup>2</sup>CV and RMSECV. The present model that used the mean center and normalization followed 211by the mean center as a pretreatment exhibited a similar level of accuracy with  $R^2CV$  of 0.787 and 212 RMSECV of 0.456. "Normalize" constitutes the pretreatment whereby the sum of the luminance is 213normalized to be 1. This pretreatment can reduce the variation caused by differences in overall brightness 214(although the spectral shapes are the same). In this study, it made no difference in precision whether 215"normalize" was performed or not; there is thus a possibility that such variation was small. The 216"autoscale" and "normalize + autoscale" were not adopted because the precision was low. 217Figure 2 (the calibration set) and Fig. 3 (the validation set) show the relationship between the cheese 218body measurement predicted from the FF using the optical fiber unit and the actually evaluated cheese 219body measurement. The prediction model was developed with three LVs using the pretreatment of the 220 mean center. In the validation set, we obtained a strong correlation ( $R^2P = 0.826$ ) and small prediction 221error (RMSEP = 0.436). Additionally, the relative standard deviation (RSD) of the body value predicted 222by FFs in the three replicates was 0.21% minimum and 5.4% maximum. The average of RSD with three 223replicates in all 30 samples was 1.6%. These values show that this method can be applied repeatedly with 224just 1-2% variation and has high reproducibility. The results shown in Figs. 2 and 3 indicate that 1) the 225FF can predict the cheese body measurement although attention is required for cheese body 226measurements of around six, and 2) using the optical fiber unit, the cheese body measurement can be 227predicted through non-contact measurement. 228The evaluation of cheese body measurements by the experts in this study often resulted in cheese 229body measurements of around six; thus, the distribution of the evaluation data was unequal. This is likely 230to have reduced the prediction accuracy. As the predicted cheese body measurements of around six have a 231higher variation than that of other cheese body measurements, the handling of results close to this value 232requires attention. To further improve the accuracy, the number of samples can be increased, or samples 233should be selected so that the distribution of cheese body measurements across the scoring range is even.

235	<i>3.3</i> .	VIP and SR scores of	of the PLSR	model plotted	using FF	contour representation

236Figure 4 shows the VIP and SR scores as a contour map to identify the range of wavelengths that 237contributes to the PLS regression model. Two wavelength ranges with high VIP scores were identified 238(Fig. 4a). Wavelengths with VIP scores of 1 or higher are known to be important variables in the model 239(Mehmood, Liland, Snipen, & Sæbø, 2012). The highest VIP score (69.9) was observed in the wavelength 240ranges with an excitation wavelength of 305 nm and an emission wavelength of 340 nm. Subsequently, the wavelength ranges with an excitation wavelength of 340 nm and an emission wavelength of 400 nm 241242also showed high VIP value (VIP score = 15.9). The VIP value of the wavelength ranges with an 243excitation wavelength of 300 nm and an emission wavelength of 680 nm was also high, but we did not 244pursue this range because it represented the second-order light of an excitation wavelength of 300 nm and 245an emission wavelength of 340 nm. As has been previously reported, the excitation wavelength of 305 nm 246and emission wavelength of 340 nm were considered to represent tryptophan and aromatic amino acids 247(Andersen et al., 2005; Andersen & Mortensen, 2008; Mazerolles et al., 2001). 248As an alternative, Fig. 4b shows a contour map of the SR scores. SR scores provide a numerical 249evaluation of the utility of each variable in a regression model. The SR scores can be used to build a 250model with even higher accuracy by excluding wavelength ranges with low SR scores and selecting 251wavelengths with only high SR scores (Rajalahti et al., 2009). In the present study, we used the F-test (95%) standard to select wavelength ranges with a large contribution (Farrés, Platikanov, Tsakovski, & 252Tauler, 2015). The SR score of the F-test (95%) was 1.55. Wavelength ranges with a SR score above this 253254value (SR score > 1.55) included the excitation wavelengths of 335–445 nm and emission wavelengths of 255370-520 nm. In these ranges, the excitation wavelength of 385 nm and emission wavelength of 460 nm 256comprised the peak wavelengths. The peak SR score for these excitation-emission wavelengths was 7.11. 257The SR score of the excitation wavelength of 350 nm and the emission wavelength of 770 nm was also 258high, but this wavelength range was also not further pursued because it represented second-order light of

the excitation wavelength of 340 nm and the emission wavelength of 400 nm.

260As VIP scores only take positive values, we used the regression coefficient (regression vector) shown 261in Fig. 4c to determine whether the wavelengths correlated positively or negatively with the cheese body 262measurements. The regression vector for each VIP peak revealed that the value was positive at the 263excitation wavelength of 305 nm and emission wavelength of 340 nm and negative at the excitation 264wavelength of 340 nm and emission wavelength of 400 nm. This suggests that the fluorescence intensity 265of the former wavelengths decreases as the cheese body measurement decreases, whereas that of the latter 266 wavelengths increases. 267Figure 5a shows the emission spectra of all samples at an excitation wavelength of 340 nm. The peak 268intensity around the emission wavelength of 400 nm increased as the cheese body measurement 269decreased; the correlation coefficient of the fluorescence at the excitation wavelength of 340 nm and 270emission wavelength of 400 nm with the cheese body measurement was -0.77. The fluorescence intensity 271of each fluorophore reflects how they are metabolized during cheese maturation. In this study, we 272considered that the wavelength range of excitation wavelength of 340 nm and emission wavelength of 400 273nm, which exhibited high correlation with cheese body measurement, was related to substances that 274increased during cheese maturation. 275Numerous investigations have focused on the range near the excitation wavelength of 340 nm and the 276emission wavelength of 400 nm. Stapelfeldt & Skibsted (1994) reported that in a model system of dairy 277products and  $\beta$ -lactoglobulin, accumulated secondary lipid oxidation products emit fluorescence at 278excitation wavelengths of 350 nm and emission wavelength of 410 nm. Morales, Romero, & 279Jiménez-Pérez (1996) monitored the fluorescence of Maillard reaction products during the thermal 280processing of a model system using milk and dairy products at an excitation wavelength of 347 nm and 281emission wavelength of 415 nm. These model system studies demonstrated that oxidized lipid and 282Maillard reaction products of milk and dairy products could be monitored via their fluorescence in the

283	range near the excitation wavelength of 340 nm and the emission wavelength of 400 nm. In addition,
284	Kokawa et al. (2015) selected peaks at an excitation wavelength of 345 nm and emission wavelength of
285	400 nm as the wavelength values with a large contribution to the PLS regression model in the FF of
286	cheese maturation indices, which suggested that oxidized lipids and Maillard reaction products are
287	present in Cheddar cheese. These are thus considered to be the same substances as measured in the
288	present study, as we also focused on these wavelengths based on the VIP values in this study. Notably,
289	although Maillard reactions do not proceed actively in hard and semi hard cheeses, studies using
290	Manchego (Corzo, Villamiel, Arias, Jiménez-Pérez, & Morales, 2000), Cheddar, Gouda and Emmental
291	(Schwietzke, Schwarzenbolz, & Henle, 2009), Harzer and Gouda cheeses (Spanneberg, Salzwedel, &
292	Glomb, 2012) have been reported that the Maillard reaction proceeded along with cheese maturation of
293	these cheeses. These studies measured Maillard reaction products in young and matured cheese using
294	high performance liquid chromatography analysis. As the amount of Maillard reaction products tends to
295	increase during cheese maturation, they are likely to show correlation with the cheese body measurement.
296	Figure 5b shows the emission spectra at an excitation wavelength of 385 nm, where the SR score was
297	the highest. The maximum wavelength of the fluorescence was between 500 and 550 nm, although
298	multiple peaks were also confirmed around 460 nm. The fluorescence peaks around 460 nm increased in
299	intensity as the cheese body measurement decreased, with the correlation coefficient between the cheese
300	body measurement and the fluorescence intensity at an excitation wavelength of 385 nm and emission
301	wavelength of 460 nm being $-0.79$ .
302	The fluorescence constituents most commonly reported in this range are the reaction products
303	between amino acids, proteins, and oxidized lipids. Kikugawa, Takayanagi, & Watanabe (1985) reported
304	that the reaction product of malondialdehyde (MDA) and lysine monomer generated by lipid oxidation
305	has a maximum excitation wavelength of 395 nm and maximum emission wavelength of 466-470 nm. In
306	addition, they reported that the reaction product of MDA and a model protein, polylysine, has a maximum

307	excitation wavelength of 398 nm and maximum emission wavelength of 470 nm. The same group
308	(Kikugawa & Beppu, 1987) reported that with the generation of fluorescent substances through lipid
309	oxidation in tissues and cells, the reaction of MDA and primary amines in the presence of monofunctional
310	aldehydes is promoted, generating fluorescent 1,4-disubstituted 1,4-dihydropyridine-3,5-dicarbaldehydes.
311	The maximum excitation wavelength of the reaction products was 386-403 nm and maximum emission
312	wavelength was 444–465 nm. Yamaki, Kato, & Kikugawa (1992) reported that the reaction product of
313	hexenal and glycine ethyl had a maximum excitation wavelength of 392 nm and maximum emission
314	wavelength of 455 nm. Veberg, Vogt, & Wold (2006) fixed the excitation wavelength at 382 nm and
315	reported the fluorescence of the reaction product of oxidized lipids (aldehydes) and amino acids. They
316	found that the reaction product of lysine and 2-hexenal has a maximum fluorescence at 471 nm, whereas
317	the reaction product of lysine and 2,4-heptadienal has a maximum fluorescence at 472 nm. The reaction
318	product of malondialdehyde and glycine shows maximum fluorescence at 465-469 nm, whereas the
319	reaction with lysine shows maximum fluorescence at 467-469 nm (both with an excitation wavelength of
320	382 nm). Furthermore, they showed that the fluorescence is generated at a temperature of 4 $^\circ$ C under
321	refrigeration. The same group (Veberg, Olsen, Nilsen, & Wold, 2007) also reported that the maximum
322	emission wavelength for oxidized lipids in butter during light irradiation was 465-470 nm, and indicated
323	that the peak emission of the oxidized lipids occurs around 470 nm regardless of the type of food (e.g.,
324	salted cod, turkey, chicken meat, or salmon pâte). In the present experiment, we did not use any
325	irradiation during storage; however, oxidized lipids may be generated during long-term storage.
326	In comparison, lumichrome, generated by the photo-oxidation of riboflavin, has an
327	excitation-emission peak near the same range; however, the peak wavelengths were an excitation
328	wavelength of 360 nm and emission wavelength of 450 nm in the model system (Fox & Thayer, 1998) but
329	an excitation wavelength of 370 nm and emission wavelength of 430 nm in yogurt (Christensen, Becker,
330	& Frederiksen, 2005). In addition to this discrepancy, the low contribution of riboflavin fluorescence to

331the model also suggests that the effect of lumichrome was small. If lumichrome had high relevance with 332the accuracy of the PLS model, riboflavin should also show high relevance. However, the wavelength 333 area corresponding to riboflavin fluorescence did not show high contribution to the PLS regression 334 model. 335Figure 5c shows the emission spectra at an excitation wavelength of 305 nm, at which there is a peak 336 in VIP. Peaks near the emission wavelength of 340 nm showed high fluorescence intensity; however, the 337 correlation with the cheese body measurement was low, with the correlation coefficient of the 338 fluorescence intensity with the cheese body measurement at an excitation wavelength of 305 nm and 339 emission wavelength of 340 nm being 0.05. Although tryptophan and aromatic amino acids represented by these peak wavelengths have been used to predict the parameters of dairy products including cheese 340 341(Andersen & Mortensen, 2008), in the present study, the correlation coefficient was low and the contribution to the PLS regression model was low. This wavelength range has strong fluorescence 342intensity and captures changes well, but by performing a comprehensive analysis such as on the FF, 343 344 changes during cheese maturation will have even higher correlation, which will allow the identification of the range of wavelengths contributing to the accuracy of the PLS regression model. We also suggest that 345346 in future studies, it should be possible to confirm the substances showing high correlation with the changes in the cheese body measurement by performing chemical analysis (e.g., extraction and high 347 performance liquid chromatography) to quantify the substances, and then comparing the relative quantity 348349 of the substance of interest with the cheese body measurement. 350

351 4. Conclusion

We showed that non-contact FF measurements using an optical fiber probe can be used to predict the cheese body measurement of Cheddar cheese. Compared with conventional fluorescence measurements in a photometer chamber using cells to hold the sample, it was possible to develop a more practical

355method. The predictive ability of the model was reliable with a strong correlation (R2P = 0.826) and small 356 prediction error (RMSEP = 0.436). Moreover, the average RSD of the three replicates in all 30 samples 357 was 1.6%. These values show that this method can be applied repeatedly with only 1-2% variation and has 358 high reproducibility. Despite the non-contact FF measurement, it was confirmed that the accuracy of this 359 prediction model is equivalent to that of previous studies. However, as the predicted values for cheese 360 body measurements of around six show large variation, it is necessary to improve the accuracy by using 361 data that reflect a more uniform distribution of cheese body measurements. 362 The wavelength ranges of oxidized lipids and Maillard reaction products (excitation wavelength of 363 340 nm and emission wavelength of 400 nm), which were assumed to make a large contribution to the 364 prediction of indices of cheese maturation (maturation time, proteolysis index, and free amino acid 365content) in previous reports, were shown to also make a large contribution to the present cheese body 366 measurement prediction model. 367 By exploring the SR scores, we discovered a new wavelength range with a large contribution to the 368 PLS regression model. These wavelengths were considered to correspond to oxidized lipids or to 369 compounds of proteins and amino acids with oxidized lipids. Additional studies are necessary to confirm 370 whether these products are responsible for these new wavelength changes, such as by comparing the 371results of quantification of these product contents with the FF measurements. Thus, our results suggest that 372the rapid and simple non-contact FF method developed herein for predicting the cheese body measurement 373 may be suitable for industrial application to enhance the quality and reliability of processed cheese 374production. 375 376 Funding: This research did not receive any specific grant from funding agencies in the public, commercial, 377 or not-for-profit sectors.

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### 379 **References**

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	fulness of	). Us	(2009)	. J. P.	Wold.	. &	5. I.	. Durán-Merás	Τ.,	. Galeano-Díaz.	D.,	odríguez.	Airado-	381
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- fluorescence excitation-emission matrices in combination with PARAFAC, as fingerprints of red
  wines. *Journal of Agricultural and Food Chemistry*, 57, 1711-1720.
- Andersen, C. M., & Mortensen, G. (2008). Fluorescence spectroscopy: a rapid tool for analyzing dairy
   products. *Journal of Agricultural and Food Chemistry*, 56, 720-729.
- 386 Andersen, C. M., Vishart, M., & Holm, V. K. (2005). Application of fluorescence spectroscopy in the
- evaluation of light-induced oxidation in cheese. *Journal of Agricultural and Food Chemistry*, 53,
  9985-9992.
- Boubellouta, T., & Dufour, E. (2008). Effects of mild heating and acidification on the molecular structure
- 390 of milk components as investigated by synchronous front-face fluorescence spectroscopy

391 coupled with parallel factor analysis. *Applied Spectroscopy*, *62*, 490-496.

- Christensen, J., Becker, E. M., & Frederiksen, C. S. (2005). Fluorescence spectroscopy and PARAFAC in
  the analysis of yogurt. *Chemometrics and Intelligent Laboratory Systems*, 75, 201-208.
- Christensen, J., Nørgaard, L., Bro, R., & Engelsen, S. B. (2006). Multivariate autofluorescence of intact
  food systems. *Chemical Reviews*, 106, 1979-1994.
- Christensen, J., Povlsen, V. T., & Sørensen, J. (2003). Application of fluorescence spectroscopy and
   chemometrics in the evaluation of processed cheese during storage. *Journal of Dairy Science, 86*,
- 398
   1101-1107.
- Corzo, N., Villamiel, M., Arias, M., Jiménez-Pérez, S., & Morales, F. J. (2000). The Maillard reaction
  during the ripening of Manchego cheese. *Food Chemistry*, 71, 255-258.
- 401 Drake, M. A., Gerard, P. D., Truong, V. D., & Daubert, C. R. (1999). Relationship between instrumental
- 402 and sensory measurements of cheese texture. *Journal of Texture Studies, 30*, 451-476.

403 Dufour, E. (2011). Recent advances in the analysis of dairy product quality using methods based on the

```
404 interactions of light with matter. International Journal of Dairy Technology, 64, 153-165.
```

- 405 Dufour, E., & Riaublanc, A. (1997). Potentiality of spectroscopic methods for the characterisation of dairy
- 406 products. i. front-face fluorescence study of raw, heated and homogenised milks. *Le Lait, 77*,
  407 657-670.
- Farrés, M., Platikanov, S., Tsakovski, S., & Tauler, R. (2015). Comparison of the variable importance in
   projection (VIP) and of the selectivity ratio (SR) methods for variable selection and

410 interpretation. Journal of Chemometrics, 29, 528-536.

- Foegeding, E. A., Brown, J., Drake, M., & Daubert, C. R. (2003). Sensory and mechanical aspects of
  cheese texture. *International Dairy Journal*, *13*, 585-591.
- Fox, J. B. Jr., & Thayer, D. W. (1998). Radical oxidation of riboflavin. *International Journal for Vitamin and Nutrition Research*, 68, 174-180.
- 415 Fujita, K., Tsuta, M., Kokawa, M., & Sugiyama, J. (2010). Detection of deoxynivalenol using
- fluorescence excitation-emission matrix. *Food and Bioprocess Technology*, *3*, 922-927.
- 417 Garimella Purna, S. K., Prow, L. A., & Metzger, L. E. (2005). Utilization of front-face fluorescence
- 418 spectroscopy for analysis of process cheese functionality. *Journal of Dairy Science*, 88, 470-477.
- 419 Guinee, T. P., Caric, M., & Kalab, M. (2004). Pasteurized processed cheese and substitute/imitation

420 cheese products. *Cheese: Chemistry, Physics and Microbiology, 2*, 349-394.

- Guinee, T. P., & Kilcawley, K. N. (2004). Cheese as an ingredient. *Cheese: Chemistry, Physics and Microbiology 2*, 395-428.
- 423 Henry, R. C., Park, E. S., & Spiegelman, C. H. (1999). Comparing a new algorithm with the classic
- 424 methods for estimating the number of factors. *Chemometrics and Intelligent Laboratory Systems*,
  425 48, 91-97.
- 426 Herbert, S., Riou, N. M., Devaux, M. F., Riaublanc, A., Bouchet, B., Gallant, D. J., et al. (2000).

- 427 Monitoring the identity and the structure of soft cheeses by fluorescence spectroscopy. *Le Lait*,
  428 80, 621-634.
- 429 ISO/IDF. (2002). Milk and milk products: Determination of nitrogen content: Routine method using
- 430 *combustion according to the Dumas principle. 185:2002.* Brussels, Belgium: International Dairy
  431 Federation.
- 432 ISO/IDF. (2004a). Cheese and processed cheese: Determination of the total solids content (Reference
  433 *method*). 4:2004. Brussels, Belgium: International Dairy Federation.
- 434 ISO/IDF. (2004b). *Cheese and processed cheese products: Determination of fat content: Gravimetric*435 *method (Reference method). 5:2004.* Brussels, Belgium: International Dairy Federation.
- 436 Kapoor, R., & Metzger, L. E. (2008). Process cheese: scientific and technological aspects a review.

437 *Comprehensive Reviews in Food Science and Food Safety, 7,* 194-214.

- Karoui, R., & Blecker, C. (2011). Fluorescence spectroscopy measurement for quality assessment of food
  systems a review. *Food and Bioprocess Technology*, *4*, 364-386.
- 440 Karoui, R., & Dufour, E. (2006). Prediction of the rheology parameters of ripened semi-hard cheeses
- using fluorescence spectra in the UV and visible ranges recorded at a young stage. *International Dairy Journal*, *16*, 1490-1497.
- 443 Karoui, R., Dufour, E., & De Baerdemaeker, J. (2006). Common components and specific weights
- 444 analysis: a tool for monitoring the molecular structure of semi-hard cheese throughout ripening.
  445 *Analytica Chimica Acta*, 572, 125-133.
- Karoui, R., Mazerolles, G., & Dufour, E. (2003). Spectroscopic techniques coupled with chemometric
  tools for structure and texture determinations in dairy products. *International Dairy Journal, 13*,
- 448 607**-**620.
- 449 Kikugawa, K., & Beppu, M. (1987). Involvement of lipid oxidation products in the formation of
- 450 fluorescent and cross-linked proteins. *Chemistry and Physics of Lipids, 44*, 277-296.

451	Kikugawa, K., Takayanagi, K., & Watanabe, S. (1985). Polylysines modified with malonaldehyde,
452	hydroperoxylinoleic acid and monofunctional aldehydes. Chemical & Pharmaceutical Bulletin,
453	33, 5437-5444.
454	Kokawa, M., Ikegami, S., Chiba, A., Koishihara, H., Trivittayasil, V., Tsuta, M., et al. (2015). Measuring
455	cheese maturation with the fluorescence fingerprint. Food Science and Technology Research, 21,
456	549-555.
457	Kulmyrzaev, A., Dufour, E., Noël, Y., Hanafi, M., Karoui, R., Qannari, E. M., et al. (2005). Investigation
458	at the molecular level of soft cheese quality and ripening by infrared and fluorescence
459	spectroscopies and chemometrics - relationships with rheology properties. International Dairy
460	Journal, 15, 669-678.
461	Kulmyrzaev, A. A., Karoui, R., De Baerdemaeker, J., & Dufour, E. (2007). Infrared and fluorescence
462	spectroscopic techniques for the determination of nutritional constituents in foods. International
463	Journal of Food Properties, 10, 299-320.
464	Lacotte, P., Gomez, F., Bardeau, F., Muller, S., Acharid, A., Quervel, X., et al. (2015). Amaltheys: A
465	fluorescence-based analyzer to assess cheese milk denatured whey proteins. Journal of Dairy
466	Science, 98, 6668-6677.
467	Lebecque, A., Laguet, A., Devaux, M. F., & Dufour, É. (2001). Delineation of the texture of salers cheese
468	by sensory analysis and physical methods. Le Lait, 81, 609-624.
469	Lee, C.H., Imoto, E.M., & Rha, C. (1978). Evaluation of cheese texture. Journal of Food Science, 43,
470	1600-1605.
471	Liu, Z., Sajith Babu, K., Coutouly, A., Allouche, F., & Amamcharla, J. K. (2016). Prediction of intact
472	casein in cheese by using Amaltheys: a front-face fluorescence analyzer. Journal of Animal
473	Science, 94, 250.
474	Mazerolles, G., Devaux, M. F., Duboz, G., Duployer, M. H., Riou, N. M., & Dufour, E. (2001). Infrared

476

and fluorescence spectroscopy for monitoring protein structure and interaction changes during cheese ripening. *Le Lait*, *81*, 509-527.

- 477 Mehmood, T., Liland, K. H., Snipen, L., & Sæbø, S. (2012). A review of variable selection methods in
- 478 partial least squares regression. *Chemometrics and Intelligent Laboratory Systems, 118*, 62-69.
- 479 Mita Mala, D., Yoshimura, M., Kawasaki, S., Tsuta, M., Kokawa, M., Trivittayasil, V., et al. (2016). Fiber
- 480 optics fluorescence fingerprint measurement for aerobic plate count prediction on sliced beef
  481 surface. *LWT Food Science and Technology*, 68, 14-20.
- 482 Mizuno, R., & Ichihashi, N. (2007). Characterization of two types of cheddar cheese as ingredients for
  483 processed cheese. *Nippon Shokuhin Kagaku Kogaku Kaishi*, 54, 395-400.
- Morales, F. J., Romero, C., & Jiménez-Pérez, S. (1996). Fluorescence associated with Maillard reaction in
  milk and milk-resembling systems. *Food Chemistry*, *57*, 423-428.
- 486 Morita, A., Araki, T., Ikegami, S., Okaue, M., Sumi, M., Ueda, R., et al. (2015). Coupled stepwise
- PLS-VIP and ANN modeling for identifying and ranking aroma components contributing to the
  palatability of cheddar cheese. *Food Science and Technology Research*, *21*, 175-186.
- 489 Muir, D. D. (2010). The grading and sensory profiling of cheese. In B. A. Law, A. Y. Tamime (Eds.).

490 *Technology of Cheesemaking* (pp. 440-474): Wiley-Blackwell.

- 491 Nakazawa, Y., & Hosono, A. (1989). *Recent advances in cheese science and technology*. New Food
  492 Industry.
- 493 Nawrocka, A., & Lamorska, J. (2013). Determination of food quality by using spectroscopic methods. In
- 494 S. Grundas & A. Stepniewski (Eds.), *Advances in Agrophysical Research* (Ch. 14). InTech.
- 495 O'Callaghan, D. J., & Guinee, T. P. (2004). Rheology and texture of cheese. In P. F. Fox, P. L. H.
- 496 McSweeney, T. M. Cogan & T. P. Guinee (Eds.), *Cheese: Chemistry, Physics and Microbiology*
- 497 (Vol. 1, pp. 511-540). London: Academic Press.
- 498 Ozbekova, Z., & Kulmyrzaev, A. (2017). Fluorescence spectroscopy as a non destructive method to

- 499 predict rheological characteristics of tilsit cheese. *Journal of Food Engineering*, 210, 42-49.
- 500 Rajalahti, T., Arneberg, R., Berven, F. S., Myhr, K. M., Ulvik, R. J., & Kvalheim, O. M. (2009).
- Biomarker discovery in mass spectral profiles by means of selectivity ratio plot. *Chemometrics* and Intelligent Laboratory Systems, 95, 35-48.
- 503 Sádecká, J., & Tóthová, J. (2007). Fluorescence spectroscopy and chemometrics in the food classification
  504 a review. *Czech Journal of Food Sciences*, 25, 159-173.
- Schwietzke, U., Schwarzenbolz, U., & Henle, T. (2009). Influence of cheese type and maturation time on
  the early Maillard reaction in cheese. *Czech Journal of Food Sciences*, 27, S140-S142.
- 507 Spanneberg, R., Salzwedel, G., & Glomb, M. A. (2012). Formation of early and advanced Maillard
- 508 reaction products correlates to the ripening of cheese. *Journal of Agricultural and Food*
- 509 *Chemistry, 60,* 600-607.
- 510 Stapelfeldt, H., & Skibsted, L. H. (1994). Modification of β–lactoglobulin by aliphatic aldehydes in
  511 aqueous solution. *Journal of Dairy Research*, *61*, 209-219.
- 512 Tamime, A. Y. (2011). Processed cheese and analogues (Vol. 16). John Wiley & Sons.
- 513 Veberg, A., Olsen, E., Nilsen, A. N., & Wold, J. P. (2007). Front-face fluorescence measurement of
- 514 photosensitizers and lipid oxidation products during the photooxidation of butter. *Journal of*
- 515 *Dairy Science*, 90, 2189-2199.
- 516 Veberg, A., Vogt, G., & Wold, J. P. (2006). Fluorescence in aldehyde model systems related to lipid
  517 oxidation. *LWT Food Science and Technology*, *39*, 562-570.
- Xiong, R., Meullenet, J. F., Hankins, J. A., & Chung, W. K. (2002). Relationship between sensory and
  instrumental hardness of commercial cheeses. *Journal of Food Science*, 67, 877-883.
- 520 Yamaki, S., Kato, T., & Kikugawa, K. (1992). Characteristics of fluorescence formed by the reaction of
- proteins with unsaturated aldehydes, possible degradation products of lipid radicals. *Chemical and Pharmaceutical Bulletin*, 40, 2138-2142.

523	Yoshimura, M., Sugiyama, J., Tsuta, M., Fujita, K., Shibata, M., Kokawa, M., et al. (2014). Prediction of

- 524 aerobic plate count on beef surface using fluorescence fingerprint. *Food and Bioprocess*
- 525 *Technology*, 7, 1496-1504.

527	Figure legends
528	
529	Fig. 1. Examples of fluorescence fingerprints (FFs) of Cheddar cheese surfaces obtained using optical
530	fiber in the range of normal intensities (0–6000 intensity (a.u.)).
531	
532	Fig. 2. Predicted vs measured cheese body measurements obtained by PLSR calibration. Bold lines show
533	$\mathbf{y} = \mathbf{x}.$
534	
535	Fig. 3. Predicted vs measured cheese body measurements obtained by PLSR validation. Bold lines show
536	$\mathbf{y} = \mathbf{x}.$
537	
538	Fig. 4. Contour plot of each score in the PLS model: (a) VIP, (b) SR, (c) regression vector.
539	
540	Fig. 5. Fluorescence emission spectra color-coded by cheese body measurement at excitation wavelengths

541 of (a) 340 nm, (b) 385 nm, and (c) 305 nm.

Age at measuments (days) Cheese Salt Storage pН Dry matter Fat Protein  $(g/100g^{-1})$  $(g/100g^{-1})$  $(g/100g^{-1})$ samples 2nd 3rd temperature (°C)  $(g/100g^{-1})$ 1st 1 231 269 318 -2 67.4 34.9 27.5 1.9 5.42 2 132 170 219 -2 5.40 67.3 34.7 27.4 2.0 3 149 -2 5.47 67.6 34.6 27.5 2.1 111 198 231 269 318 5 5.48 67.7 34.9 27.3 1.9 4 132 170 219 5 5.43 67.9 35.3 27.7 2.2 5 269 318 10 5.39 67.7 35.5 27.2 1.9 6 231 7 203 241 290 10 5.52 67.6 35.5 27.4 1.9 8 231 269 318 15 5.55 68.1 27.2 2.1 35.6 9 203 241 290 15 5.47 68.0 35.1 27.7 2.1 2.1 10 132 170 219 15 5.41 67.6 35.3 27.5

Table 1Physico-chemical compositions of the 10 Cheddar cheeses.

Pretreatment	LVs	RMSECV	R <sup>2</sup> CV	$R^2C *$	$R^2P$	RMCEP
mean center	3	0.456	0.787	0.800	0.826	0.436
normalize + mean center	3	0.456	0.787	0.808	0.747	0.515
autoscale	3	0.493	0.757	0.822	0.787	0.470
normalize + autoscale	2	0.482	0.762	0.787	0.739	0.519

 Table 2

 Results of PLS 1

 egression for four pretreatment methods

\* The coefficients of determination for calibration.

