

Identification of a freshness marker metabolite in stored soybean sprouts by comprehensive mass-spectrometric analysis of carbonyl compounds

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Abstract: The objective of this study was to identify metabolites that quantitatively indicate degrees of freshness of soybean sprouts. Self-cultivated soybean sprouts were stored at 5 °C, 10 °C or 20 °C, and respiratory CO₂ production rates were monitored using a gas chromatography during storage. Carbonyl compounds (CCs) were analyzed comprehensively using mass-spectroscopic metabolomics analyses. CCs were derivatized using dansyl hydrazine (DH) and were then analyzed using high performance liquid chromatography-electrospray ionization tandem mass spectrometry (HPLC-MS/MS) with multiplexed multiple reactions monitoring (MRM). In the MRM chromatogram, 171 to 358 peaks were observed from stored soybean sprouts. Principle component analysis and discriminant analysis (PCA-DA) selected CC-DH derivative ion with m/z 512 at a retention time of 9.34 min as the most significant metabolite. Searching online metabolomics database and matching fragment patterns of product ion mass spectra of an authentic standard revealed abscisic acid is a freshness marker of soybean sprouts.

1 **Identification of freshness marker metabolite in stored soybean sprouts by**
2 **comprehensive mass-spectrometric analysis of carbonyl compounds**

3

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26 **ABSTRACT**

27 The objective of this study was to identify metabolites that quantitatively indicate degrees of
28 freshness of soybean sprouts. Self-cultivated soybean sprouts were stored at 5 °C, 10 °C or 20
29 °C, and respiratory CO₂ production rates were monitored using a gas chromatography during
30 storage. Carbonyl compounds (CCs) were analyzed comprehensively using mass-
31 spectroscopic metabolomics analyses. CCs were derivatized using dansyl hydrazine (DH) and
32 were then analyzed using high performance liquid chromatography-electrospray ionization
33 tandem mass spectrometry (HPLC-MS/MS) with multiplexed multiple reactions monitoring
34 (MRM). In the MRM chromatogram, 171 to 358 peaks were observed from stored soybean
35 sprouts. Principle component analysis and discriminant analysis (PCA-DA) selected CC-DH
36 derivative ion with m/z 512 at a retention time of 9.34 min as the most significant metabolite.
37 Searching online metabolomics database and matching fragment patterns of product ion mass
38 spectra of an authentic standard revealed abscisic acid is a freshness marker of soybean
39 sprouts.

40

41 *Keywords:* Abscisic acid, Carbonyl compound, Freshness assessment, Metabolomics,
42 Soybean sprouts.

43

44 **1. Introduction**

45 Fresh fruits and vegetables are critical components of human diets and provide
46 many health benefits (Slavin & Lloyd, 2012). Although freshness is associated with
47 attractiveness and nutritional benefit, most consumers do not have direct access to fresh fruits
48 and vegetables, and it can take several days for distribution of commercial harvests to
49 consumers. Because substantial losses of nutritional value can occur during distribution,
50 freshness of fruit and vegetable may facilitate consumer's decisions to purchase produce, and

51 could also be used to suggest appropriate postharvest techniques for farmers, distributors and
52 retailers who intend to maintain freshness from farm to table.

53 Although freshness of fresh fruits and vegetables is widely assessed using
54 observations of the color change and degree of wilting of produces, validated assessments of
55 freshness are often required to avoid purchasing substandard produce, because visual
56 assessment is limited to the skill of person. In particular, early reductions in freshness are
57 often not visible (Nilsson, 2000).

58 Fruit and vegetable deterioration reflects increased prevalence of senescence (Paliyath
59 & Droillard, 1992). Plant senescence is a complex and highly regulated process that is
60 characterized by the degradation of chlorophyll, carotenoid, protein, and cell membrane and
61 loss of moisture (Biswal, 1995; Zhou & Gan, 2009). Among these, cellular membrane
62 integrity is an important indicator (Dörnenburg & Davies, 1999). In most cases, cell
63 membrane degradation follows lipid decomposition and is indicated by increasing
64 peroxidized lipid contents (Paliyath & Droillard, 1992), decreasing phospholipids (Lester &
65 Whitaker, 1996) and unsaturated fatty acid levels (Lester, 2003). Numerous peroxidized
66 lipids including subsequent production of carbonyl compounds (CCs) such as aldehydes,
67 ketones, and carboxylic acids (Wills, 1980) are formed during lipid degradation. Therefore,
68 specific CC accumulates during membrane lipid degradation as well as senescence could be a
69 potential biomarker for quantitative freshness assessment of fruits and vegetables.

70 Recently, metabolomics using analytical instrumentations such as gas
71 chromatography (GC) and liquid chromatography (LC) coupled with mass spectrometry
72 (MS) has been introduced as a powerful approach to identify biomarkers in postharvest
73 science (Singh, 2015). For instance, by means of GC-MS based metabolomics, 10 volatile
74 compounds have been identified as potential markers of chilling injury of basil leaves
75 (Cozzolino et al., 2016). Pentane and 2-ethylfuran have been detected as markers of quality

76 changes of stored wild rocket (Luca, Kjær & Edelenbos, 2017). In addition, Rudell, Mattheis
77 and Hertog (2009) employed untargeted metabolic profiling to characterize metabolomic
78 changes associated with superficial scald development in “Granny Smith” apple following 1-
79 methylcyclopropene or diphenylamine treatment. In case of tomato fruits, LC-MS based
80 metabolomics has been introduced to characterize metabolic changes during ripening (Moco
81 et al., 2006). However, in the case of freshness biomarker identification, LC-MS based
82 metabolomics is not yet done. Since, Tomono, Miyoshi and Ohshima (2015) developed and
83 validated a method for detecting trace levels of CCs in mice plasma using HPLC/ESI-MS/MS
84 with multiplexed multiple reactions monitoring after derivatization by dansyl hydrazine
85 (DH), we applied this method for the establishment of CC profiles in fruits and vegetables to
86 identify the freshness maker.

87 Soybean sprouts are a popular vegetable globally, and especially in Japan, Korea,
88 China, and other Southeast Asian countries (Huang, Cai & Xu, 2014). Soybean sprouts are
89 rich in nutrients and are available all year round, but are highly perishable due to high
90 respiration rate (Snowdon, 2010). Therefore, a quantitative freshness assessment is necessary
91 for postharvest management of soybean sprouts.

92 Brash, Charles, Wright and Bycroft (1995) indicated that rate of fruits and vegetables
93 perishability is related to the cumulative respiratory CO₂ production during postharvest.
94 Therefore, the cumulative postharvest CO₂ production can be used as a reference indicator for
95 the degree of freshness. Hence, we compared CCs contents with cumulative CO₂ production
96 during storage of soybean sprouts under various temperature conditions, and selected specific
97 CC as a freshness biomarker. Subsequently, we identified the structure of the specific CC by
98 using online metabolomics databases and confirmed by comparing fragmentation of mass
99 spectra with the authentic standard.

100

101 **2. Materials and methods**

102 *2.1. Plant material*

103 Soybean sprouts were produced using *Glycine max*, cv. BS5012 seeds.

104

105 *2.2. Reagents*

106 Abscisic Acid (ABA), DH and *p*-Toluenesulfonic acid (*p*-TsOH) were purchased
107 from Sigma-Aldrich (St. Louis, MO, USA). *p*-Benzyloxybenzaldehyde (*p*-BOBA),
108 acetonitrile, methanol, chloroform and formic acid were obtained from Wako Pure Chemical
109 Industries (Osaka, Japan). Butylated Hydroxytoluene (BHT) was purchased from Nacalai
110 Tesque (Kyoto, Japan).

111

112 *2.3. Cultivation and storage conditions*

113 Ten g samples of soybean seeds were sterilized by dipping into water at 70 °C for 10 s
114 and were then incubated (MIR-154-PJ Panasonic, Gunma, Japan) for 8 h in water at 20 °C to
115 induce the germination. After soaking, the seeds were placed in 250 mL plastic cups, which
116 were used as cultivation chambers, and were incubated under the dark condition at 20 °C with
117 70 % – 80 % relative humidity (RH). The seeds were watered with 100 mL of tap water twice
118 daily at 10 am and 4 pm. After 4 days cultivated sprouts were harvested and selected on the
119 basis of uniform hypocotyl lengths of 10 ± 2 cm. Samples were divided into three groups and
120 each was stored in the incubators set at 5 °C, 10 °C and 20 °C with 70 % – 80 % RH and were
121 collected periodically for measurements.

122

123 *2.4. Measurement of respiration rate by a flow-through method*

124 Rates of respiratory CO₂ production of stored soybean sprouts were measured by a
 125 flow-through method using on-line gas chromatography (GC) as described in Fahmy and
 126 Nakano (2014) with some modifications. Briefly, 40 g of soybean sprouts was placed into an
 127 acrylic chamber (2 L) equipped with gas inlet and outlet tubes. The chambers were closed
 128 and fresh air was flowed into the chamber from an air compressor through the inlet tubes at
 129 the flow rate of 6 L h⁻¹. The chambers were then placed in incubators at 5 °C, 10 °C and 20
 130 °C. Inlet and outlet gas samples were injected automatically into a GC (GC-14A Shimadzu,
 131 Kyoto, Japan) alternately via a 0.5 mL sampling loop attached to a rotating stepping valve.
 132 CO₂ was separated using a Porapak Q column and detected by a thermal conductivity
 133 detector. Helium gas was used as a carrier gas. The chromatograms were analyzed using an
 134 integrator (C-R7A plus Shimadzu, Kyoto, Japan) based on a CO₂ standard curve. The results
 135 were expressed as percentage of total gas volume. The rate of CO₂ production was calculated
 136 from the differences in gas concentration between the inlet and outlet using Eq. (1) (Fonseca
 137 et al. 2002).

$$138 \quad \left(\frac{C_2 - C_1}{C_1} \right) \frac{1}{1 - \frac{C_2}{C_1}} = \frac{R}{P} \frac{W}{F} \frac{1}{T} \quad (1)$$

139 where R is the respiration rate for CO₂ production (mmol kg⁻¹ h⁻¹), C_1 and C_2 are
 140 volumetric concentration of CO₂ in inlet and outlet gas samples, respectively (%), W is the
 141 weight of the sample (kg), F is flow rate (L h⁻¹), P is the atmospheric pressure (= 101.3 kPa),
 142 R is the universal gas constant (= 8.314 L kPa K⁻¹ mol⁻¹) and T is the absolute temperature
 143 (K).

144 CO₂ production rates of soybean sprouts were monitored every hour during storage
 145 and cumulative CO₂ production after harvest was calculated by integrating respiration rates
 146 throughout the storage duration using the trapezoidal rule.

147

148 *2.5. Sample preparation and CCs extraction*

149 Fleshly harvested soybean sprouts and those stored for 4, 8 and 12 d at 5 °C, 2, 4 and
150 6 d at 10 °C, and 0.5, 1 and 2 d at 20 °C were collected for determination of CCs contents.
151 Collected sprouts were then divided into cotyledon and hypocotyl parts and about 100 mg
152 samples of precisely weighted cotyledon and hypocotyl were placed in 2 ml self-standing
153 screw cap microtubes (Watson, Kobe, Japan) containing single zirconia ball of 5 mm
154 diameter. Samples were soaked into liquid nitrogen for 2 min and were finally stored at -80
155 °C for further analysis.

156 Frozen sprout samples were crushed in 200- μ L aliquots of methanol containing
157 0.05% BHT using a bead crusher (Shake Master Neo BMS, Tokyo, Japan) at 1500 rpm for
158 180 s. Subsequently, 100- μ L aliquots of 0.1 μ mol mL⁻¹ *p*-BOBA were added as an internal
159 standard. After adding 400- μ L aliquots of chloroform, mixtures were homogenized again
160 using the bead crusher for 120 s, and mixtures were then centrifuged at 12000 rpm for 10 min
161 at 10 °C (1720 Kubota, Osaka, Japan). Finally, organic phases were collected and were
162 derivatized with DH.

163

164 *2.6. DH derivatization*

165 DH derivatization was performed as described by Tomono et al. (2015) with some
166 modifications. Briefly, 200- μ L organic phases were mixed with 400- μ L aliquots of
167 acetonitrile containing 200 μ g of DH and 40 μ g of *p*-TsOH using a water bath shaker
168 (Personal 11 Taitec, Saitama, Japan) at 75 rpm for 4 h in the dark at 30 °C. Mixtures were
169 then evaporated to be complete dried using a centrifugal evaporator (CV-2100 Eyela, Tokyo,
170 Japan) and residues were then dissolved in 500 μ L of acetonitrile and filtered through 0.2- μ m
171 membranes (RC15 Minisart Sartorius, Göttingen, Germany). Finally, 5- μ L aliquots were
172 analyzed using LC/ESI-MS/MS.

173

174 2.7. HPLC/ESI-MS/MS analysis

175 CCs were determined using a HPLC series system with a high pressure gradient
176 pump, an autosampler, a column oven (Prominence HPLC 20A Shimadzu, Kyoto, Japan),
177 and a reverse phase chromatographic column (Unison UK-C8, 150 mm × 2.0 mm i.d., 3 μm
178 in particle size, Imtakt, Kyoto, Japan) coupled to a triple-quadrupole mass spectrometer (Q-
179 TRAP 4500 AB-Sciex, Framingham, MA, USA). Elution was performed in binary gradient
180 mode with mobile phases comprising 0.1 % formic acid in water (solvent A) and 0.1% formic
181 acid in acetonitrile (solvent B). DH-derivatized CCs (CC-DHs) were ionized using a Turbo-
182 VTM ion source in positive mode and were detected using the multiplexing MRM of a specific
183 product ion with an *m/z* value of 236.1 by collision-induced dissociation. A total of 400
184 MRM transitions were monitored for all CC-DHs and a total of 100 channels were monitored
185 simultaneously for each sample injection. One channel from each injection was reserved for
186 monitoring of the transition of *p*-BOBA-DH as internal standard (IS) at *m/z* 460–236.1.
187 Determinations were performed in three replications with five injections for each sample to
188 complete multiplexing of the 400 MRM transitions. The detector conditions were as follows:
189 ion-spray voltage, 5500 V; source temperature, 300 °C; curtain gas, 206.8 kPa; collision gas,
190 62.1 kPa; ion source gas 1 (sheath gas), 344.7 kPa; ion source gas 2 (drying gas), 551.6 kPa.
191 Declustering potentials and collision energies were 100 V and 37 eV for *m/z* 275–374, 115 V
192 and 39 eV for *m/z* 375–474, 120 V and 43 eV for *m/z* 475–574, and 130 V and 50 eV for *m/z*
193 575–674, respectively. Nitrogen was used as a collision gas. To determine structures of
194 candidate freshness marker metabolites, the selected CC-DH ion was fragmented in the
195 product ion scanning detection mode.

196

197 2.8. Data processing

198 Data from peaks of CC derivatives were processed using Marker View™ software
199 1.2.1. (AB-Sciex, Framingham, MA, USA). CC-DH peaks were extracted using Gaussian
200 smoothing of 1.5 points, noise percentage of 50%, a baseline subtraction window of 8 min, a
201 peak splitting factor of 4 points, a retention time tolerance of 1 min, a minimum intensity of
202 1500 cps, a minimum peak with 2 points, and a minimal signal/noise ratio of 20. Peak areas
203 were normalized to that of the IS and to sample weights. Principal component analysis with
204 discriminant analysis (PCA-DA) was then performed using Pareto scaling and none
205 weighting in order to find out any differential features between sample groups. Selected
206 marker ion candidates were then identified using The Lipidmaps (www.lipidmaps.org), The
207 Human Metabolome Database (www.hmdb.ca), The METLIN Metabolomic Database
208 (<https://metlin.scripps.edu>), The ChemicalBook Database (www.chemicalbook.com) and
209 compared with a purchased authentic standard.

210

211 **3. Results and discussion**

212 *3.1 Change of CO₂ production rate of soybean sprouts during storage*

213 Figure 1 shows the changes of CO₂ production rate of soybean sprouts during storage
214 at 5 °C, 10 °C, and 20 °C. CO₂ production rate of soybeans sprouts stored at 5 °C was almost
215 stable between 0.8 – 1.4 mmol kg⁻¹ h⁻¹ during storage, while it was slightly decreased from
216 2.4 to 1.5 mmol kg⁻¹ h⁻¹ at 10 °C. On the other hand, CO₂ production rate at 20 °C was
217 drastically decreased from 9 to 5 mmol kg⁻¹ h⁻¹ during storage. The rate of respiration is
218 affected by not only temperature but also the amount of the time elapsed after harvest.
219 According to Brash et al. (1995), the respiration rate of asparagus stored at 20 °C decreased
220 by over 50% in the first 24 hours after harvest, and the decrement was less significant with
221 decreasing of temperatures. Deterioration of fresh produce is primarily driven by the
222 product's own tissue metabolism and there is a tight linkage between metabolism and

223 perishability. Since CO₂ productions provide parallel measures of metabolic activity, the
224 cumulative CO₂ production could be used as a reference of degree of freshness. In latter
225 sections, we discuss the relationship between the cumulative CO₂ production and change of
226 CCs in soybean sprouts during storage to identify the potential freshness marker.

227

228 3.2 Profile of CCs in soybean sprouts

229 Figure 2 demonstrates CC metabolites features in fresh soybean sprouts (A; cotyledons
230 and B; hypocotyls). Detected CC-DHs were plotted in circles as a function of retention time
231 (RT) and *m/z* value. The diameter of each circle represents peak area of detected CC-DH
232 normalized to that of IS-DH at a RT of 11.34 min and to respective sample weight. Even in
233 the fresh condition, about 171 of CC-DHs in cotyledons (A) and 228 of CC-DHs in
234 hypocotyls (B) were detected. Most of these CC-DHs were distributed at the range of
235 retention time from 3 min to 15 min and *m/z* from 350 to 600. Many of CCs in fresh soybean
236 sprouts have a wide range of polarity and molecular weight that might be considered as
237 secondary metabolites because there are numerous CCs classified as secondary metabolites
238 including vitamins, isoflavones, flavonols, chalcones, and their derivatives which have
239 carbonyl skeletons (Di Carlo, Mascolo, Izzo, & Capasso, 1999; Kim, Kim, Chung, Chi, Kim,
240 & Chung, 2006; Gu et al., 2017). They likely react with DH forming CC-DH derivatives after
241 extraction in polar and semi-polar solvent mixtures thus leading to the detection of numerous
242 CC-DHs. Moreover, secondary metabolites in soybean sprouts vary in hypocotyls and
243 cotyledons depending on the soybean variety (Plaza, Ancos & Cano, 2003; Youn, Kim, Lee
244 & Kim, 2011). From our data, we can assume that, in soybean sprouts of *Glycine max*, cv.
245 BS5012, higher number of CCs species are distributed in hypocotyls compared to cotyledons.

246 Figure 3 demonstrates the score plots (A) and corresponding loading plots of CC-DH
247 signals (B) from hypocotyls of stored soybean sprouts relating to cumulative CO₂ production.

248 Four sample clusters were circled and positioned in different areas of the score plot to
249 discriminate differences between groups as a function of cumulative CO₂ production, which
250 was observed in first, second, third, and fourth clusters at 0, 87–110, 155–208, 274–299
251 mmol kg⁻¹, respectively, and increased with D1 scores (Fig. 3A). However, distributions of
252 CC-DHs in cotyledons were not distinguished relating to the cumulative CO₂ production in
253 PCA-DA score plots (data not shown). From corresponding loading plot of CC-DH signals
254 from hypocotyls (Fig. 3B), increases in cumulative CO₂ production are explained by the
255 positive direction of D1 where three signals were circled at the positive edge of D1 axis of
256 the loading plot that responsible for clustering of the samples and had *m/z* of Q1_*m/z* of
257 Q3_RT pairs of 364_236.1_8.82, 512_236.1_9.34, and 330_236.1_17.29.

258 Figure 4 demonstrates the correlation coefficient (*r*) of each detected CC-DH signal
259 that was extracted by regression analysis between each of its normalized peak area and
260 cumulative CO₂ production during storage under various temperatures. The three signals
261 chosen previously from PCA-DA results indicate positive relationship with cumulative CO₂
262 production where signal of 512_236.1_9.34 has the highest *r* value (*r* = 0.84) (Fig. 4A).
263 Moreover, accumulation of the CC-DH signal 512_236.1_9.34 shows an increasing trend
264 with the increase of cumulative CO₂ production during storage under various temperatures in
265 regression analysis (*R*² = 0.71) (Fig. 4B). Therefore, only the signal of 512_236.1_9.34
266 representing CC-DH metabolite with *m/z* 512 at RT of 9.34 min was finally selected as a
267 freshness marker ion of soybean sprouts.

268

269 3.3 Structure elucidation of freshness marker ion

270 The enhance product ion (EPI) scan detection in Analyst[®] system of AB-SCIEX was
271 performed against the selected ion of *m/z* 512 with RT of 9.34 min to identify the selected
272 freshness marker ion. Figure 5 demonstrates the product ions mass spectra of the selected ion

273 of m/z 512 at RT of 9.34 min (upper part) and that of the standard abscisic acid (ABA)-DH
274 (lower part). Initially, the losses of fragment ions with m/z of 18 mass units were observed in
275 Fig. 5 (upper part). It likely reflects the loss of water molecules from protonated hydroxyl
276 groups in allylic positions (Britton, 1996) and suggesting the presence of hydroxyl ions.
277 Generally, a hydroxyl ion can be derived from fatty acids, carotenoids, flavonoids and their
278 conjugates. Furthermore, since the selected marker ion is a protonated ion molecule, $[M+H]^+$,
279 it comprises CC, DH and H^+ (hydrogen-adduct ion). Therefore, to calculate the molecular
280 weight (MW) of the CC from this CC-DH derivative, m/z values of 1 for hydrogen-adduct
281 ions and 263 for DH moieties were subtracted from the detected m/z value of 512, and an m/z
282 value of 16 was added for the atomic mass of oxygen to form a carbonyl skeleton (Fig. S1).
283 Based on these calculations, the MW of 264 mass units for the selected freshness marker
284 metabolite was extracted. Using all the information obtained in our analysis, we have
285 searched the online metabolomics databases for the corresponding names and structures of
286 possible candidates. Five compounds were nominated as candidates, and are listed with their
287 formulas, structures, and hydrophobicity ($\log P$) values in Table 1. These compounds are
288 characterized as derivatives of fatty acids, flavonoids, and their conjugates. In addition, since
289 a reverse phase chromatographic separation system was used for separating the complex CC-
290 DHs in the samples, the resulting RTs reflect the polarities. Specifically, the RT of the
291 selected CC-DH derivative was 9.34 min and was faster than that of *p*-BOBA-DH, which was
292 detected at 11.34 min as an IS-DH derivative. Therefore, the selected freshness marker ion is
293 more polar than *p*-BOBA-DH, and from the compounds listed with $\log P$ values in Table 1,
294 only ABA has a lower $\log P$ value than *p*-BOBA. Thus, to confirm that ABA is the present
295 freshness marker metabolite in soybean sprouts, we purchased an authentic ABA standard
296 and conducted EPI detection against the ABA-DH derivative (Fig. 5, lower part), and
297 indicated that fragmentation patterns of product ion mass spectra of ABA-DH exactly

298 matched those of the selected CC-DH. In addition, the RT of the ABA-DH derivative was
299 9.41 min, similar to that of the selected CC-DH. Finally we conclude that ABA is the
300 identified freshness marker metabolite for soybean sprouts.

301 ABA is a plant hormone that was discovered at least 50 years ago and has since been
302 shown to regulate many aspects of plant growth and development (Finkelstein, 2013). ABA
303 is a member of monocyclic monoterpene family and comprises the metabolic precursors
304 ketone and enolate (Duffield & Netting, 2001). Thus, ABA can be conjugated to DH through
305 its ketone group. To date, the best known functions of ABA are related to roles as a major
306 phytohormone that contributes to plant abiotic stress resistance. ABA is mainly induced by
307 moisture loss stress, chilling temperature and salt stress (Swamy & Smith, 1999; Lafuente &
308 Sala, 2002; Romero, Rodrigo, & Lafuente, 2013), and accumulates through the cleavage of a
309 C₄₀ carotenoid precursor (Xiong & Zhu, 2003). According to Becker and Fricke (1996), fresh
310 fruits and vegetables lose their moisture through the transpiration during storage. Hence, the
311 presence of ABA in hypocotyls of soybean sprouts may have been induced by moisture loss.
312 Transpiration is associated with transport and evaporation of moisture from the skin, and with
313 convective mass transport of moisture to the atmosphere. Moreover, transpiration and
314 respiration have been correlated in previous study whereas CO₂ and heat from the associated
315 chemical reaction during respiration may accelerate transpiration in fresh produce.

316 As indicated in Fig. 4B, normalized areas of ABA signals at 512_236.1_9.34,
317 increased with cumulative CO₂ production. However, this correlation was moderate ($R^2 =$
318 0.71). As mentioned in the introduction part, senescence can be characterized by some
319 indications such as degradation of chlorophyll, protein and lipid as well as moisture loss.
320 Thus, we suggest that in combination with ABA levels, other prospective freshness marker
321 metabolites may give more valid assessments of freshness of soybean sprouts.

322 The present experiments did not show accumulation of specific lipid degradation
323 derived CCs, potentially reflecting limitations of our analytical methods. Although DH
324 derivatization-based LC-MS is satisfactory for profiling of CCs in biological samples, the
325 formation of by-products (artifacts) and non-quantitative reactions can hamper accurate
326 determinations (Moritz & Johansson, 2008; Xu, Zou, Liu, Zhang, & Ong, 2011; Qi, Liu,
327 Wang, Cai, Yuan, & Feng, 2014). Moreover, plant materials such as soybean sprouts contain
328 large varieties of metabolites with carbonyl skeletons that could lead to an abundance of DH
329 derivative artifacts and non-quantitative reactions. Hence, the present conditions may impair
330 detection efficacy for inherently low levels of lipid degradation derived CCs in test samples.
331 These limitations warrant further optimization of the present method to minimize interference
332 from complex biological matrices and to enhance detection selectivity.

333

334 **4. Conclusion**

335 Herein, the first use of HPLC–MS/MS based metabolomics approach to identify
336 markers of freshness in stored soybean sprouts was reported. ABA was identified as a
337 metabolite that can indicate the degree of soybean sprouts freshness. Although ABA has been
338 associated previously with responses to abiotic stresses such as moisture loss, no studies
339 suggest the use of ABA as a metabolite biomarker for freshness. Hence, the present data are
340 the first to suggest the utility of ABA as a marker for freshness of soybean sprouts,
341 particularly because ABA was absent in freshly harvested sprouts and accumulated during
342 storage. However, these data advise further validation of using ABA as a marker of freshness
343 in soybean sprouts, as well as in other fruits and vegetables.

344

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349

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449

450 **Figures captions**

451 **Fig. 1** CO₂ production rates of soybean sprouts stored at various temperatures.

452 **Fig. 2** CC metabolites feature in fresh soybean sprouts; cotyledon (A) and hypocotyl (B).

453 **Fig. 3** Score scatter plot (A) and loading scatter plot (B) of the PCA-DA of CC-DHs in the
454 hypocotyls in stored soybean sprouts. Values beside data points in Fig. 3A indicate
455 cumulative CO₂ production during storage (mmol kg⁻¹ FW). Values beside data points
456 in Fig. 3B indicate *m/z* of Q1_ *m/z* of Q3_RT pairs for each CC-DH signal.

Fig. 4 Relationship between each normalized peak area of CC-DH signal and cumulative
CO₂ production during storage under various temperatures; (A) coefficient correlation
of each CC-DH and *m/z* of CC-DHs. (B) Cumulative CO₂ production and linear
regression of CC-DHs with *m/z* 512 at RT of 9.34 min and cumulative CO₂
production.

Fig. 5 Product ions mass spectra of protonated molecules ion with *m/z* of 512 at RT of
9.34 min in positive ion mode; (A) selected freshness marker ion and (B) authentic
ABA standard-DH.

FIGURE 1

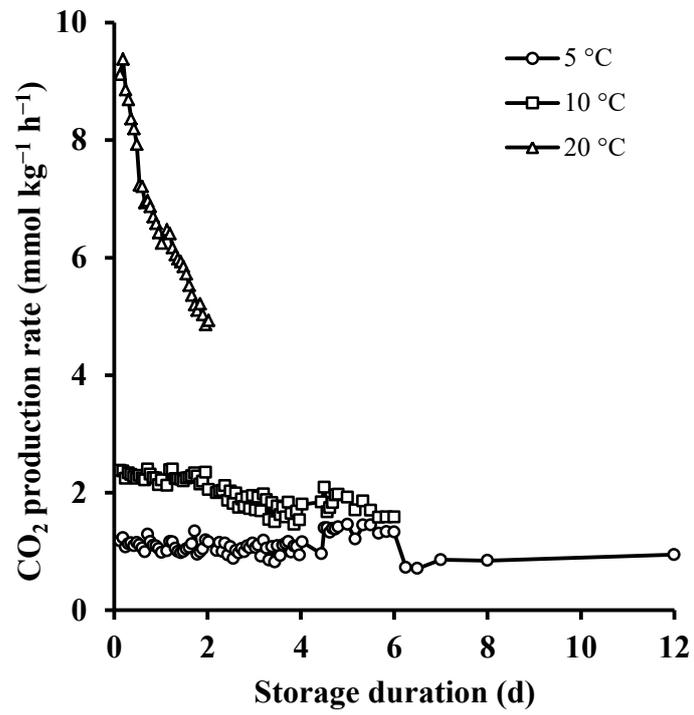
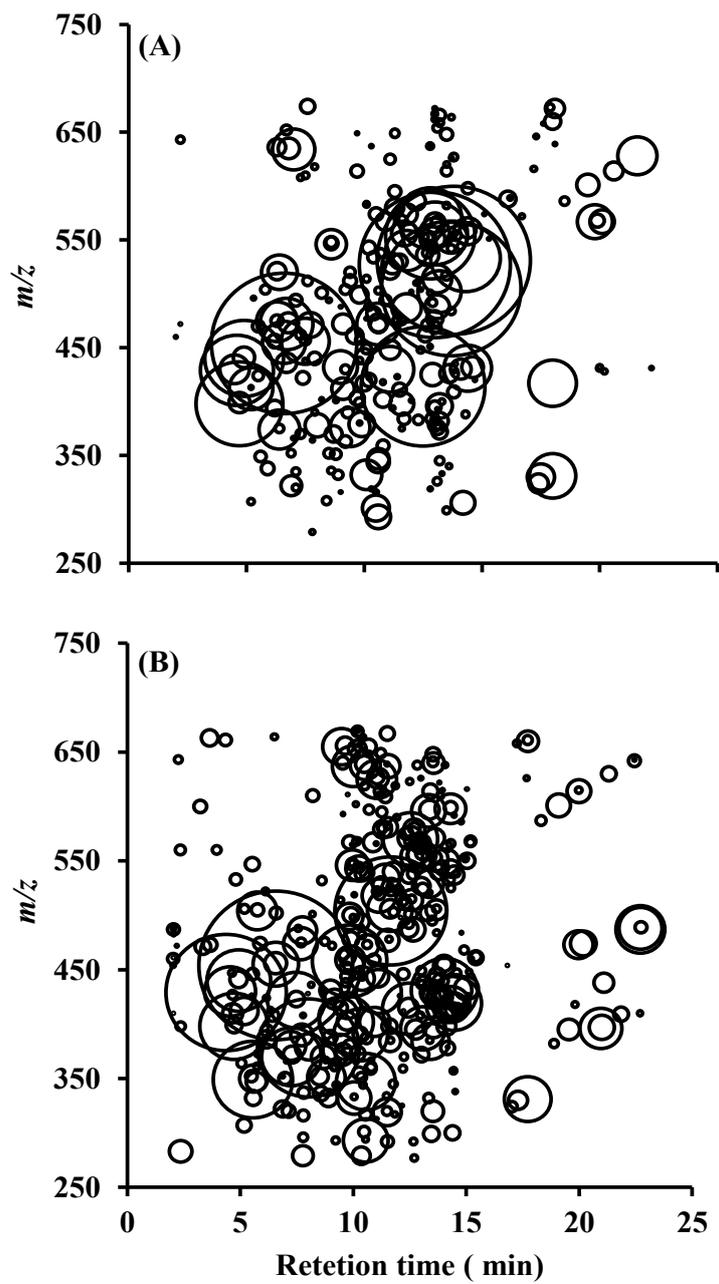
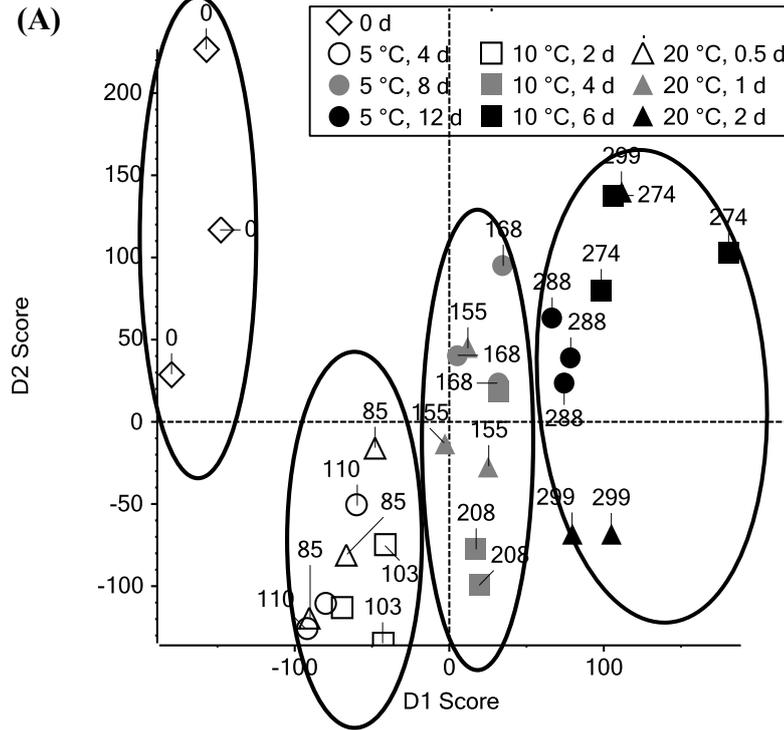


FIGURE 2



Scores for D1 (37.8 %) versus D2 (36.2 %), Pareto (DA)



Loadings for D1 (37.8 %) versus D2 (36.2 %), Pareto (DA)

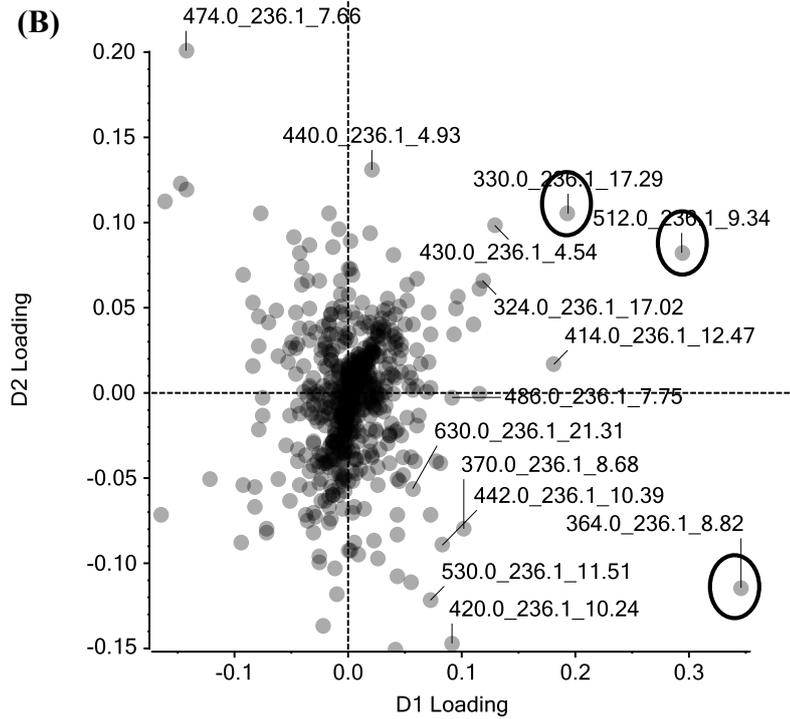


FIGURE 4

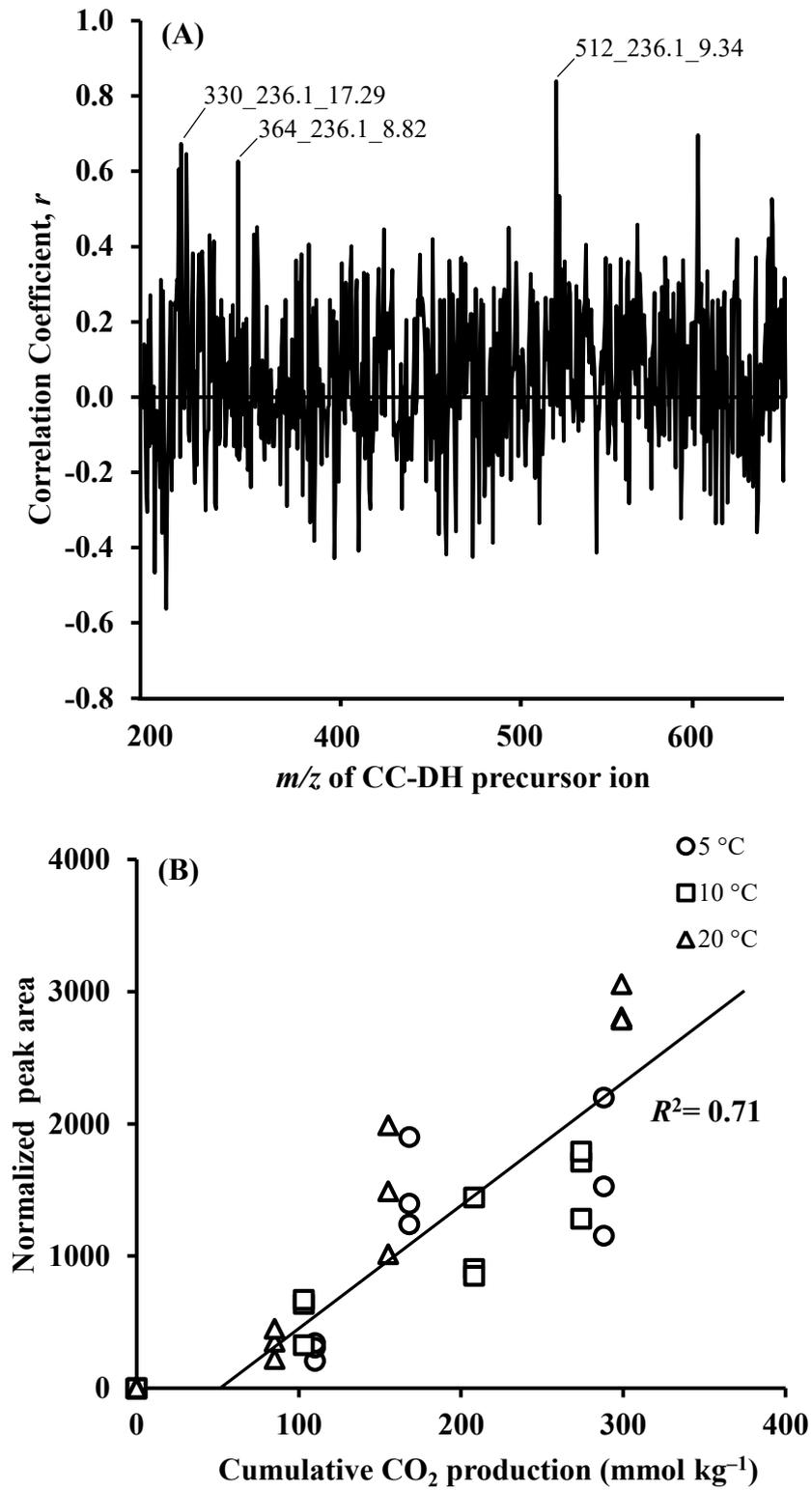


FIGURE 5

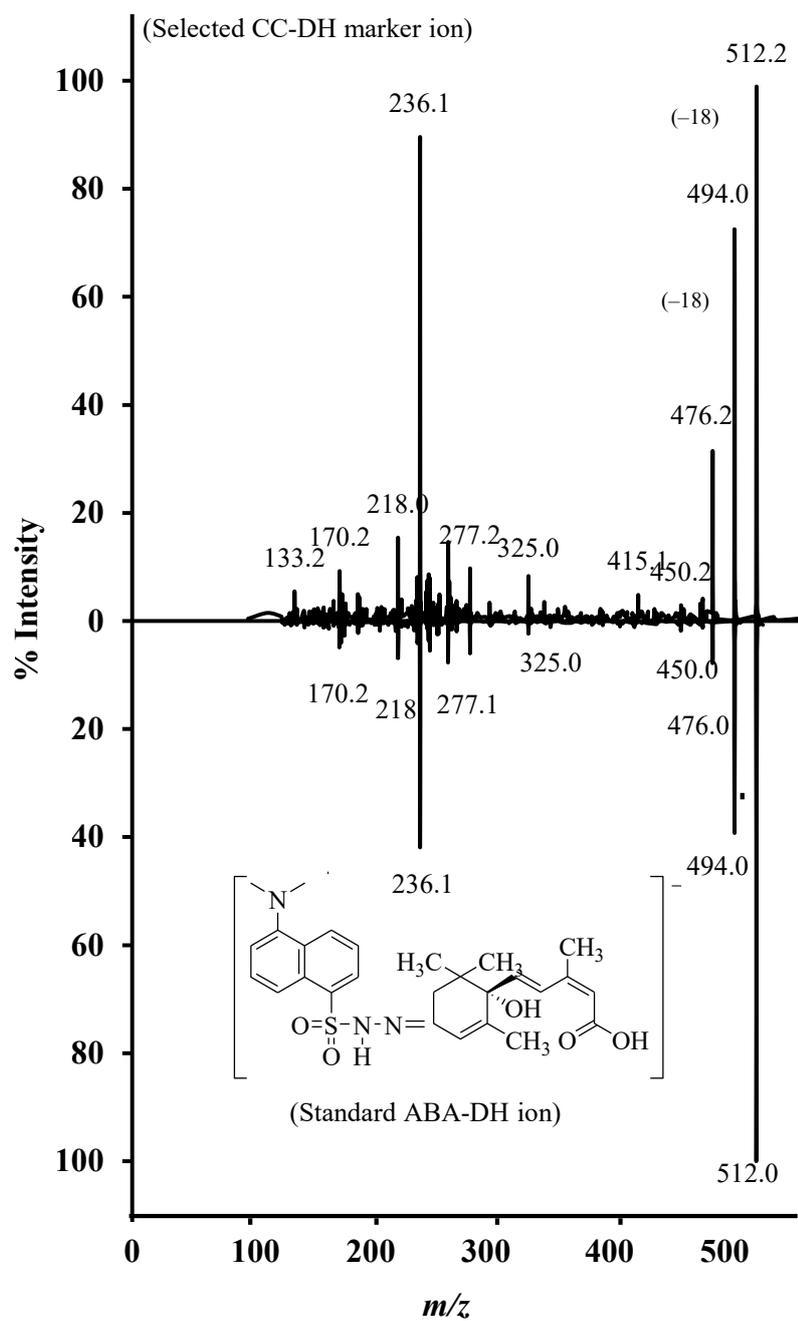
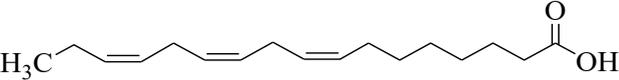
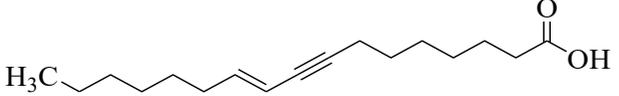
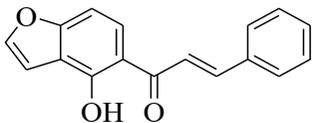
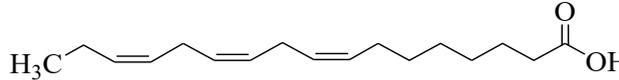
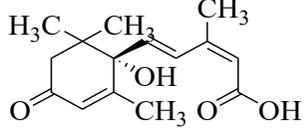
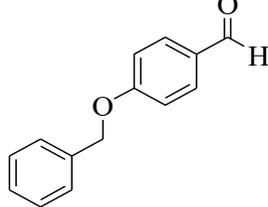


Table 1 List of the name for the compound with MW of 264 that contain hydroxyl group (loss of $m/z=18$) and internal standard (RT of 11.4 min)

No	Name	Formula	Log P (Hydrophobicity)	Structure	On line databases
1	Norlinolenic acid	$C_{17}H_{28}O_2$	5.27		www.lipidmaps.org
2	10E-heptadecen-8-ynoic acid	$C_{17}H_{28}O_2$	4.94		www.lipidmaps.org
3	2'-Hydroxyfurano [2'',3'':4',3']chalcone	$C_{17}H_{12}O_3$	4.03		www.lipidmaps.org
4	all-trans-7-hydroxy hexadeca-2,4,8,10-tetraenoic acid	$C_{16}H_{24}O_3$	3.91		www.lipidmaps.org
5	Abscisic acid	$C_{15}H_{20}O_4$	2.54		www.lipidmaps.org
6	<i>p</i> -BOBA (IS)	$C_{14}H_{12}O_2$	3.3		www.chemicalbook.com

Supplementary Material

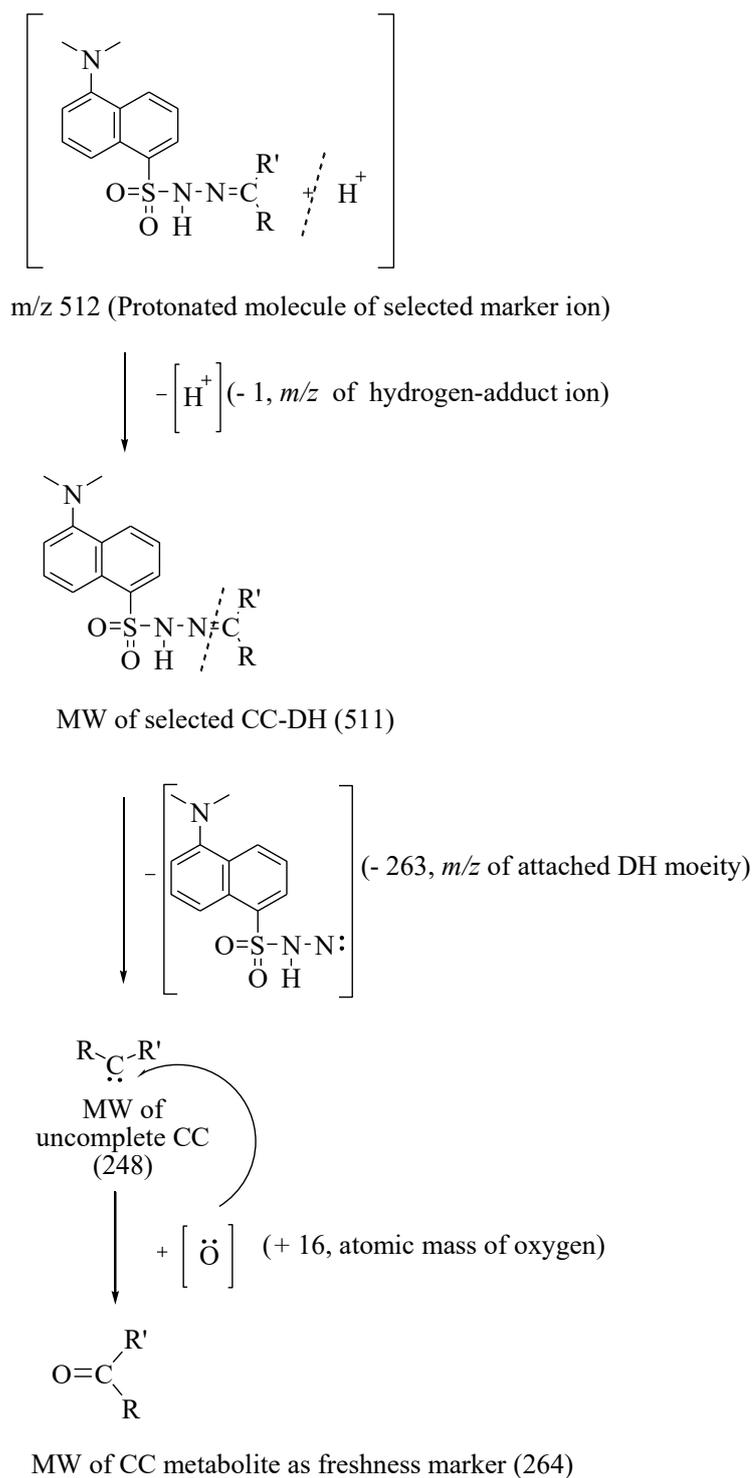


Fig. S1. Schematic of methods for determining the molecular weight of the CC-DH freshness marker metabolite of soybean sprouts with an m/z of 512 and a RT of 9.34 min.